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CLAIMS

1. A homogeneous process for the preparation of reactive sol-gel porous materials either chemically or physically doped with stable organic nitroxyl radicals, comprising the steps of:

a) compolymerizing a solution including:

at least one monomer precursor selected from the group including metal and semi-metal alkoxides, metal esters or semi-metal esters, of the general formula

 $M(R)_{D}(P)_{B}$

wherein M is a metal or a semimetal. R is an hydrolyzable substituent, P is a non-polymerizable group, n is an integer of 1 to 6, and m is an integer of 0 to 6,

a dopant consisting of a stable di-tertiary-alkyl nitroxyl radical or a precursor thereof of formula

wherein A represents a chain of two or threewcarbon atoms, one or two of said carbon atoms being eventually substituted by one oxygen or nitrogen atom,

a solvent including H₂O and a co-solvent generally selected from aliphatic alcohols;

an acid or base to catalyse the processes of solgel hydrolysis and copolymerization; and

one or more additives selected from those known to be useful in the preparation of porous materials

to form a gel containing said dopant trapped therein;

- b) evaporating said solvent;
- c) drying said gel;

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d) coating said gel on a mesoporous inorganic support; and

e) drying said mesoporous material coated with said sol-gel.

2. The process according to claim 1, wherein M is Si, Ti, Pb or Al and R is an alkowy or arylalkowy group.

3. The process according to claim 1 or 2, wherein said non-hydrolyzable substituent is H, an alkyl, aryl or fluoroalkyl group or an aminoalkyl group.

4. The process according to any one of the preceding claims, wherein said step c) of drying the gel is a liophilisation carried out at a pressure lower than 70 mm Hg and preferably 15 mm Hg, to obtain a mesoporous aerogel powder.

preceding claims, wherein said step c) of drying the gel is a mild heat treatment carried out at an atmospheric pressure and a temperature lower than 100° C, and preferably at 45° C.

6. The process according to claim 1, wherein said nitroxyl radical is added to said solution along with said monomer precursor in a one-step procedure.

7. The process according to claim 1, wherein in a two-step procedure, first said monomer precursor is hydrolyzed in part with water in the presence of an acid and then said nitroxyl radical is added to this solution, to obtain a porous sol-gel polymeric oxide with a fractal macromolecular structure.

8. The process according to any one of the preceding claims, wherein said nitroxyl radical is 4-oxy-TEMPO and said monomer precursor is 3-aminopropyl-trimethoxysilane to obtain a catalytic material containing chemically linked radicals.

9. The process according to claim 8, wherein said radical is tethered to said monomer precursor through reductive amination by stirring for three hours a solution of 4-oxo-TEMPO in methanol with a slight excess

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of 3-aminopropyl-trimethoxysilane, and reducing the thus formed imine with NaBH3CN.

10. The process according to any one of claims 1 to whereta said mitroxyl radical is TEMPO or a precursor thereof and said radical is physically entrapped within a sol-gel matrix adding a solution thereof in methanol to said precursor following either said one- or two-step procedure as claimed in claim 6 or 7.

11. The process according to any one of the preceding claims, wherein said catalytic porous materials are in the shape of powders, films, monoliths, or fibers.

12. Use of a doped porous material obtained with a process as claimed in any one of claims 1 to 11, to catalytically mediate a liquid-phase oxidative conversion of primary and secondary alcohols into carbonyl or carboxyl darivatives in the presence of a primary oxidant.

13, A process for the liquid-phase oxidative conversion of a substrate of a primary or secondary alcohol into a carbonyl or carboxyl derivative thereof, characterized in that said oxidative conversion is carried out in the presence of a doped catalytic material as defined in any one of claims 1 to 11, and in the presence of a primary oxidant effective in selective alcohol oxidations mediated by nitroxyl radicals.

14. A process according to claim 13, wherein said liquid phase is as organic solvent, a biphasic organic solvent-water system, or water and said primary oxidant is NaOCl, NaOBr, HNO; CuCl/O2, K3Fe(CN); or NO3

15. A process according to claim 13 or 14, wherein said alcohol substrate is an alkyl alcohol, an aryl alcohol, a steroid alcohol, an allylic alcohol, a terpenoid alcohol or retinol and it is oxidated in a biphasic reaction system CH2Cl2-H2O, said primary oxidant is aqueous alkaline NaOSI and said catalytic material is prepared according to chaim 8 or 9 and it is added to the reaction system under stirking.

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- 16. A process according to claim 13 or 14, wherein said alcohol substrate is a monomer or an oligomeric carbohydrate protected at the anomeric center, solvent is water, said exident is alkaline NaOCl or NaOCl in the presence of a catalytic amount of NaBr, and said cayalytic material added to the reaction solution is one as claimed in anyone of claims 8, 9 or 10.
- 17. A process\according to claim 16, wherein said catalytic material is in the form of pumice stones coated with a sol-gel film doped with said ntroxyl radicals as claimed in anyone of claims 1 to 10 and said carbohydrate is a water soluble polymer.
- 18. The catalytic material doped with a chemically linked nitroxyl radical obtained with a process as claimed in anyone of claims 8 or 9.

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